

# Alkalinity (Hydroxyl) Test Kit

1 drop = 10 or 50 ppm as  $\text{CaCO}_3$  / 25 mL

# TK1026-Z

red caps

## KIT COMPONENTS:

HA6094-B	Hydrochloric Acid 0.12N, 60 mL
HA6117-B	Hydrochloric Acid 0.6N, 60 mL
BC2013-B	Barium Chloride 20%, 60 mL
PH1605-A	Phenolphthalein Indicator, 30 mL
SY-2001-P	Syringe, 1 mL
VL-1005-V	Vial, 10-50 mL

**INTERFERENCES:** Turbid samples may mask the color change at the endpoint. Use a pH meter for these samples.

## SAFETY TIPS:



Wear  
Gloves



Use Eye  
Protection



Read  
SDS

## TESTING TIPS:



Collect  
Accurate  
Sample



Hold  
Bottles  
Vertically



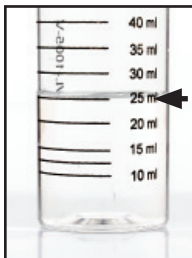
Ensure  
Proper  
Lighting

**ATTENTION:** As necessary, calibrate this kit against a known standard made with plant / make-up water. Be sure to collect a representative sample.

Video Procedure



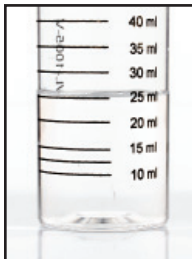
**1** Rinse vial three times with sample to be tested. **Fill vial to 25 mL.**



STEP 1

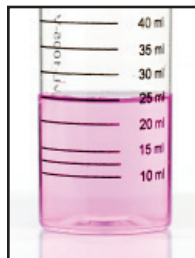
**2** Add 20 drops of Barium Chloride 20% (BC2013) and swirl to mix.

*Note: You may also remove the dropper tip and use the syringe to add 1 mL of Barium Chloride 20%.*



STEP 2

**3** Add 2 drops of Phenolphthalein Indicator (PH1605) and swirl to mix. The sample will turn pink if alkalinity is present.

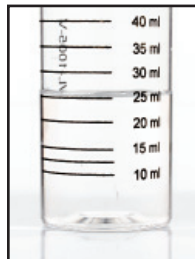


STEP 3

**4** Add Hydrochloric Acid Titrant one drop at a time while swirling. Count the number of drops until the sample color changes from pink to colorless.

Hydrochloric Acid 0.12N (HA6094)  
# drops x 10 = ppm as  $\text{CaCO}_3$

Hydrochloric Acid 0.6N (HA6117)  
# drops x 50 = ppm as  $\text{CaCO}_3$



STEP 4