

Hydrogen Peroxide Test Kit

1 drop = 50 ppm as H_2O_2 / 10 mL

TK3340-Z

green caps

KIT COMPONENTS:

ND7505-B	Peroxide Titrant, 60 mL
FE3144-A	Ferriin Indicator, 30 mL
SY-2010-P	Syringe, 10 mL
VL-1005-V	Vial, 10-50 mL

SAFETY TIPS:



Wear
Gloves



Use Eye
Protection



Read
SDS

TESTING TIPS:



Collect
Accurate
Sample



Hold
Bottles
Vertically



Ensure
Proper
Lighting

INTERFERENCES: This method is affected by any oxidizable substances in the sample such as organic matter, Sulfides, Hydrogen Sulfide, and mercaptans. If present, these substances will interfere by reacting with the titrant, yielding an erroneously high Hydrogen Peroxide concentration. Iron and Lead ions can cause precipitation. Cupric and Ferrous ions cause low results.

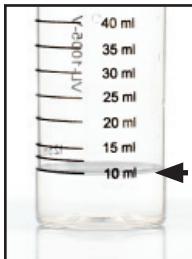
ATTENTION: As necessary, calibrate this kit against a known standard made with plant / make-up water. Be sure to collect a representative sample.



Video Procedure

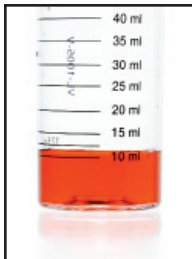


1 Rinse vial three times with sample to be tested. **Fill vial to 10 mL.**



STEP 1

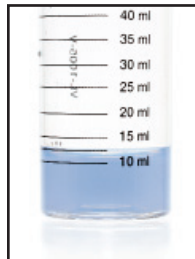
2 Add **8 drops of Ferriin Indicator** (FE3144) and swirl to mix. The sample should turn orange.



STEP 2

3 Add **Peroxide Titrant** (ND7505) one drop at a time while swirling. Count the number of drops until the sample color changes from orange to blue – blue / green.

$$\# \text{ drops} \times 50 = \text{ppm as H}_2\text{O}_2$$



STEP 3

Hydrogen Peroxide (HR) Test Kit

1 drop = 500 ppm as H_2O_2 / 1 mL

TK3340-Z

green caps

KIT COMPONENTS:

ND7505-B	Peroxide Titrant, 60 mL
FE3144-A	Ferriin Indicator, 30 mL
SY-2010-P	Syringe, 10 mL
VL-1005-V	Vial, 10-50 mL

SAFETY TIPS:



Wear
Gloves



Use Eye
Protection



Read
SDS

TESTING TIPS:



Collect
Accurate
Sample



Hold
Bottles
Vertically



Ensure
Proper
Lighting

INTERFERENCES: This method is affected by any oxidizable substances in the sample such as organic matter, Sulfides, Hydrogen Sulfide, and mercaptans. If present, these substances will interfere by reacting with the titrant, yielding an erroneously high Hydrogen Peroxide concentration. Iron and Lead ions can cause precipitation. Cupric and Ferrous ions cause low results.

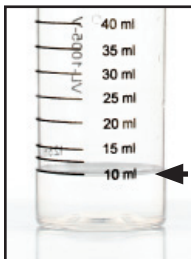
ATTENTION: As necessary, calibrate this kit against a known standard made with plant / make-up water. Be sure to collect a representative sample.



Video Procedure

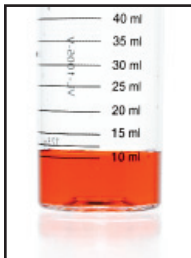


1 Rinse vial three times with sample to be tested. Use the syringe to **collect 1 mL sample and add to the vial.** Then, **fill vial to 10 mL** with water.



STEP 1

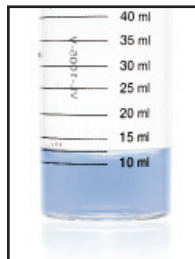
2 Add **8 drops of Ferriin Indicator (FE3144)** and swirl to mix. The sample should turn orange.



STEP 2

3 Add Peroxide Titrant (ND7505) one drop at a time while swirling. Count the number of drops until the sample color changes from orange to blue – blue / green.

$$\# \text{ drops} \times 500 = \text{ppm as H}_2\text{O}_2$$



STEP 3